## Penicillin-Cephalosporin conversion iii. A novel route to 3-chloromethyl- $\Delta^3$ -cephems

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ABSTRACT: Synthesis of 3-chloromethyl- $\triangle^3$ -cephems 3 from 4-arenesulfonylthioazetidin-2-ones 1, derived from penicillins G and V, has been achieved by the electrolytic ene-type chlorination of 1 in a CHCl<sub>3</sub>-aqueous NaCl-H<sub>2</sub>SO<sub>4</sub>-(Pt electrodes) system and subsequent ring closure with NH<sub>3</sub> in DMF.

3-Chloromethyl- $\Delta^3$ -cephems 3 are important precursors in the synthesis of 3'-substituted cephalosporin antibiotics. 1) They have been prepared from 3-acetoxymethyl cephalosporins by displacement of the acetoxy group with clorine atom. 2,3) Recent growing interest in the penicillin-cephalosporin conversion as an economical sense enable us to develop a straightforward synthetic route to 3 from azetidinones 1, prepared from natural penicillins. 4)

The conversion of  $\underline{1}$  to  $\underline{3}$  comprises the electrolytic ene-type chlorination<sup>5)</sup> of  $\underline{1}$  and the ring closure of  $\underline{2}$  with base (Scheme 1). We found that the arenesulfonyl groups (Ar-SO<sub>2</sub>) have a sufficient nature for both protecting the thiol groups at C(4)-position in electrolysis conditions ( $\underline{1} \longrightarrow \underline{2}$ ) and playing the part of leaving groups in cyclization conditions ( $\underline{2} \longrightarrow \underline{3}$ ). Some of our results are summarized in the Table.

A typical procedure is as follows (entry 1). A solution of  $\underline{la}$  (R<sup>1</sup> = PhCH<sub>2</sub>, R<sup>2</sup> = CH<sub>3</sub>, Ar = p-NO<sub>2</sub>Ph, 0.2 mmol) in CHCl<sub>3</sub> (5 ml) and aqueous NaCl (2 g/6 ml) containing H<sub>2</sub>SO<sub>4</sub> (0.14 ml)

## Scheme 1

$$R^{1} \stackrel{O}{C}NH$$
  $S-SO_{2}Ar$   $R^{1} \stackrel{O}{C}NH$   $S-SO_{2}Ar$   $NH_{3}$   $R^{1} \stackrel{O}{C}NH$   $S-SO_{2}Ar$   $NH_{3}$   $R^{1} \stackrel{O}{C}NH$   $S-SO_{2}Ar$   $NH_{3}$   $S-SO_{2}Ar$   $S-SO_{2}Ar$ 

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Table.				yıelds, % <sup>a)</sup>	
entry	R <sup>1</sup>	R <sup>2</sup>	Ar	<u>1</u> →2	$\underline{2} \rightarrow \underline{3}$
1	PhCH <sub>2</sub>	CH <sub>3</sub>	p-NO <sub>2</sub> Ph	83	74
2	PhCH <sub>2</sub>	CH 3	Ph	<i>7</i> 7	82
3	PhCH <sub>2</sub>	PhCH <sub>2</sub>	p-NO <sub>2</sub> Ph	91	86
4	PhCH <sub>2</sub>	PhCH <sub>2</sub>	Ph	84	78
5	PhCH <sub>2</sub>	p-NO <sub>2</sub> PhCH <sub>2</sub>	p-NO <sub>2</sub> Ph	75	52
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a) Isolated yields after column chromatography (SiO<sub>2</sub>).

PhCH<sub>2</sub>

were charged in a cell fitted with two Pt foil electrodes (1.5 x 2 cm<sup>2</sup>). Electrolysis was carried out under a constant current of 10 mA/cm<sup>2</sup> at 7-12 °C for 1.3 h (8 F/mol of electricity passed). The organic layer was separated and worked up in the usual manner, yielding  $\frac{2a}{2}$  (R<sup>1</sup> = PhCH<sub>2</sub>, R<sup>2</sup> = CH<sub>3</sub>, Ar = p-NO<sub>2</sub>Ph, 83%): IR (CHCl<sub>3</sub>) 3400 (NH), 1780, 1745, 1673 cm<sup>-1</sup> (C=0);  $\frac{1}{1}$ H NMR (CDCl<sub>3</sub>) 8 3.57 (s, 2H), 3.76 (s, 3H), 4.17 (bs, 2H), 4.90 (s, 1H), 5.05 (dd, 4.4, 7 Hz, 1H), 5.18 (s, 1H), 5.35 (s, 1H), 5.94 (d, 4.4 Hz, 1H), 6.50 (d, 7 Hz, 1H), 7.28 (s, 5H), 7.98 (d, 8 Hz, 2H), 8.37 (d, 8 Hz, 2H).

p-NO<sub>2</sub>Ph

94

93

The ring closure of  $\underline{2a}$  (0.1 mmol) was accomplished by treatment with NH  $_3$  (gas, 0.3 mmol) in DMF (0.5 ml) at -20  $\sim$  -30 °C for 1 h, yielding  $\underline{3a}$  (R<sup>1</sup> = PbCH<sub>2</sub>, R<sup>2</sup> = Me, 74%) without contamination of the  $\triangle^2$ -isomer. With regard to the cyclization with base, use of gaseous ammonia was found to be most effective among the following bases (yields of  $\underline{3a}$ ): Acona (29%) Et N (18%); KOH (17%); and KI (14%).

## References

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